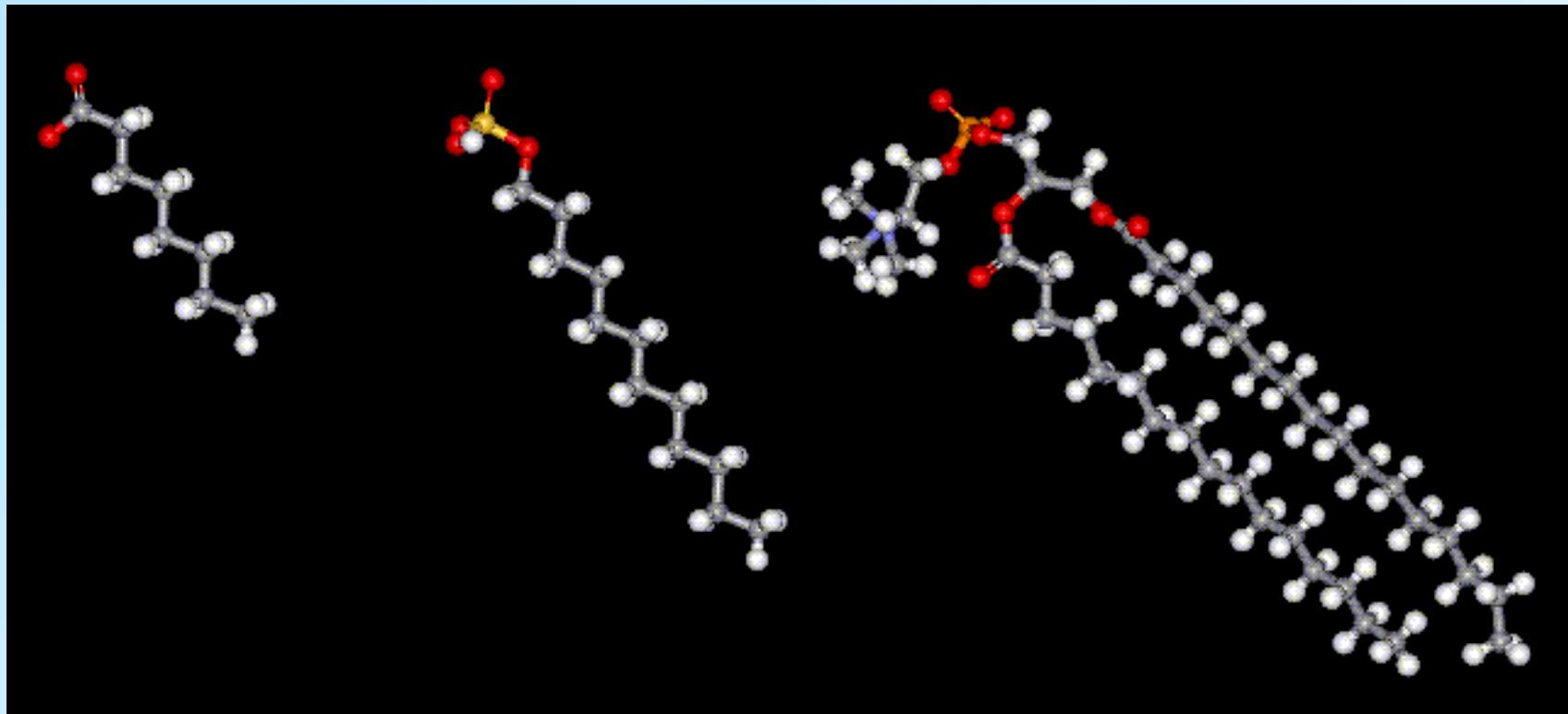
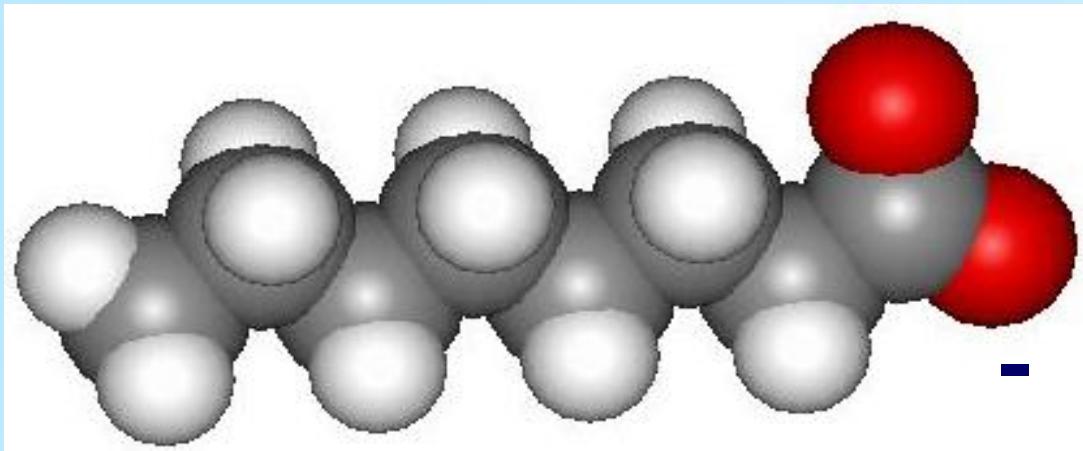


# *Molécules amphiphiles*



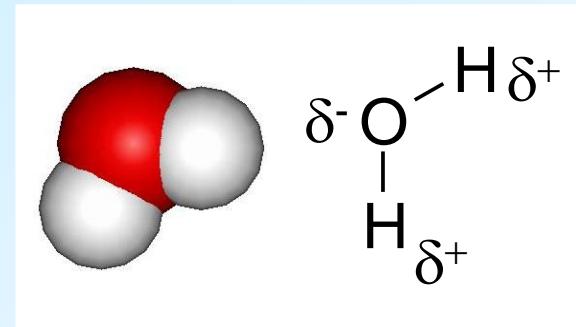
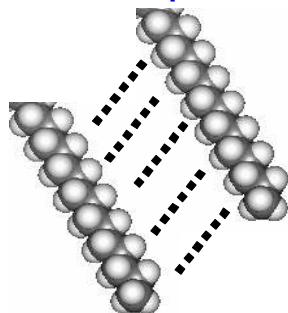


lipophile

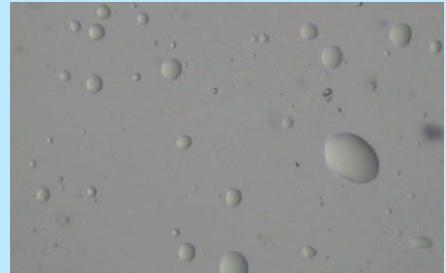
hydrophile

### Chaînes carbonées apolaires

Elles interagissent avec  
d'autres chaînes carbonées  
forces de Van der Waals  
entre dipôles électriques temporaires

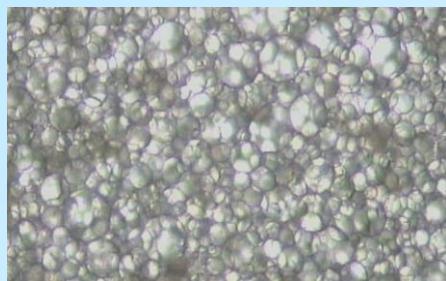


# *Mayonnaise*



*Gouttes d'huile en suspension dans l'eau*

*mayonnaise :  
état 1*



*mayonnaise :  
état 2*

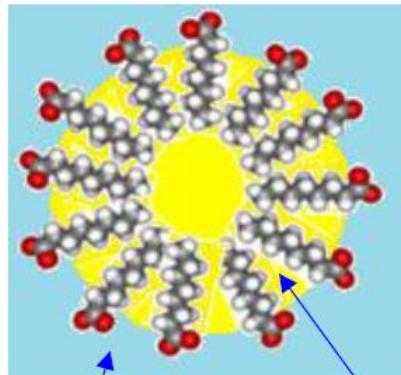


*mayonnaise :  
état 3*

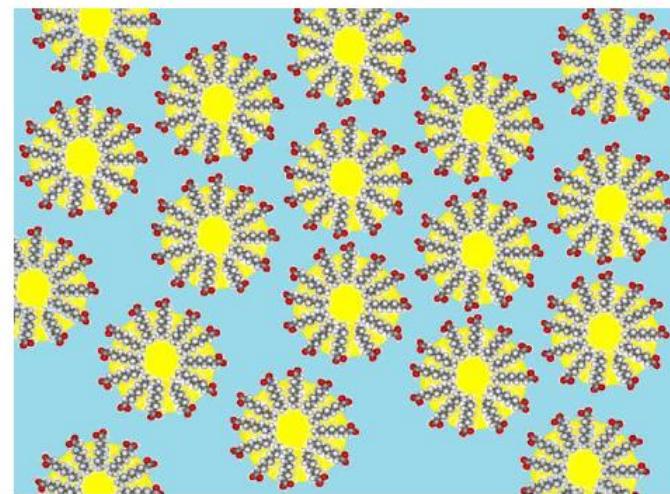


# MICELLE

La mayonnaise est une **émulsion** constituée de gouttes d'huile, **les micelles**, séparées par un film d'eau.



Eau



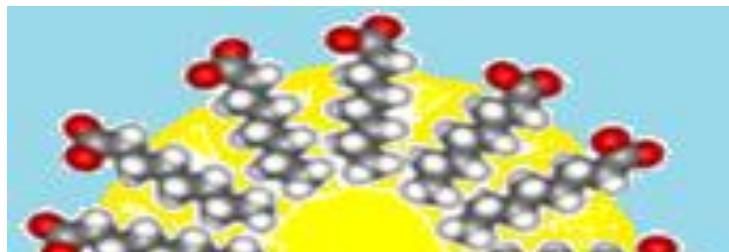
Huile

# Nettoyage

Tache d'huile

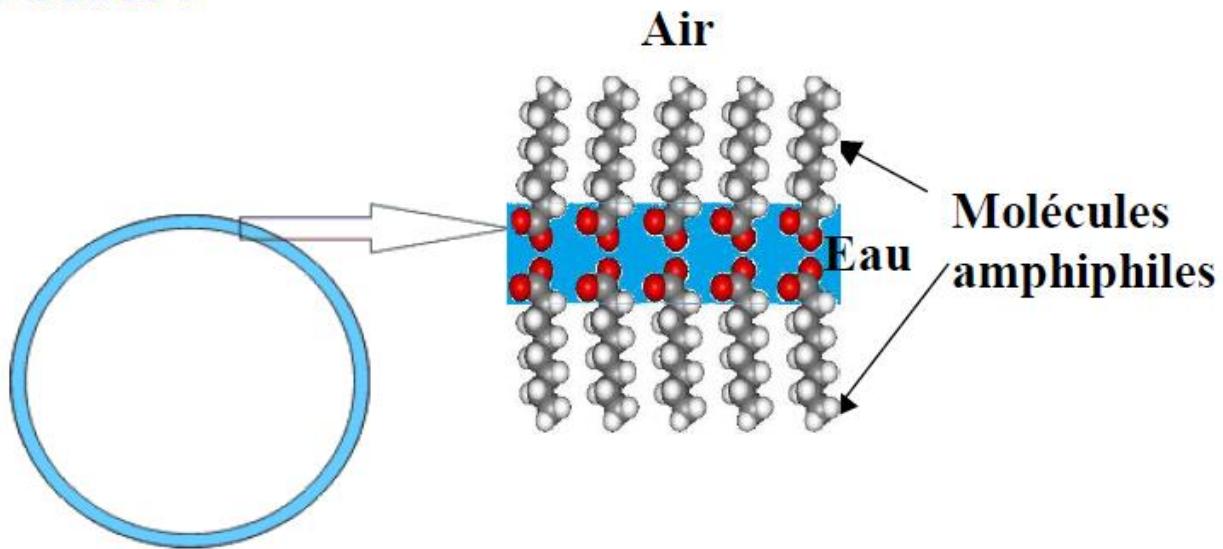
Tissu

Eau + détergeant

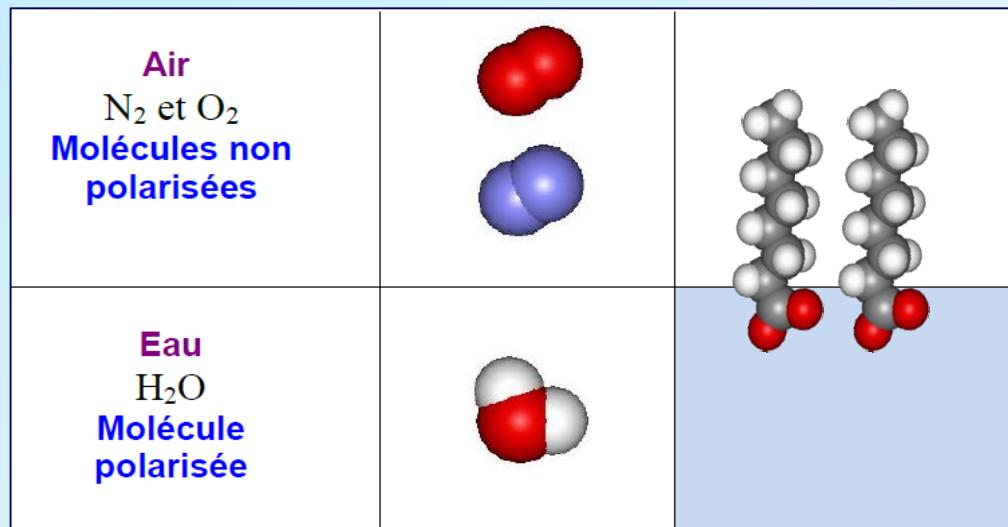
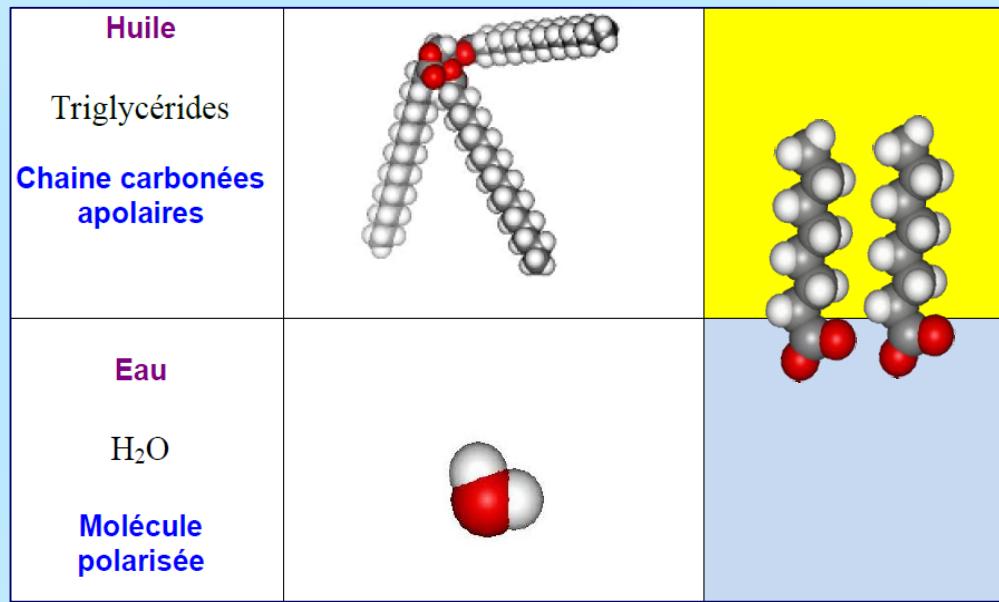


# Bulles

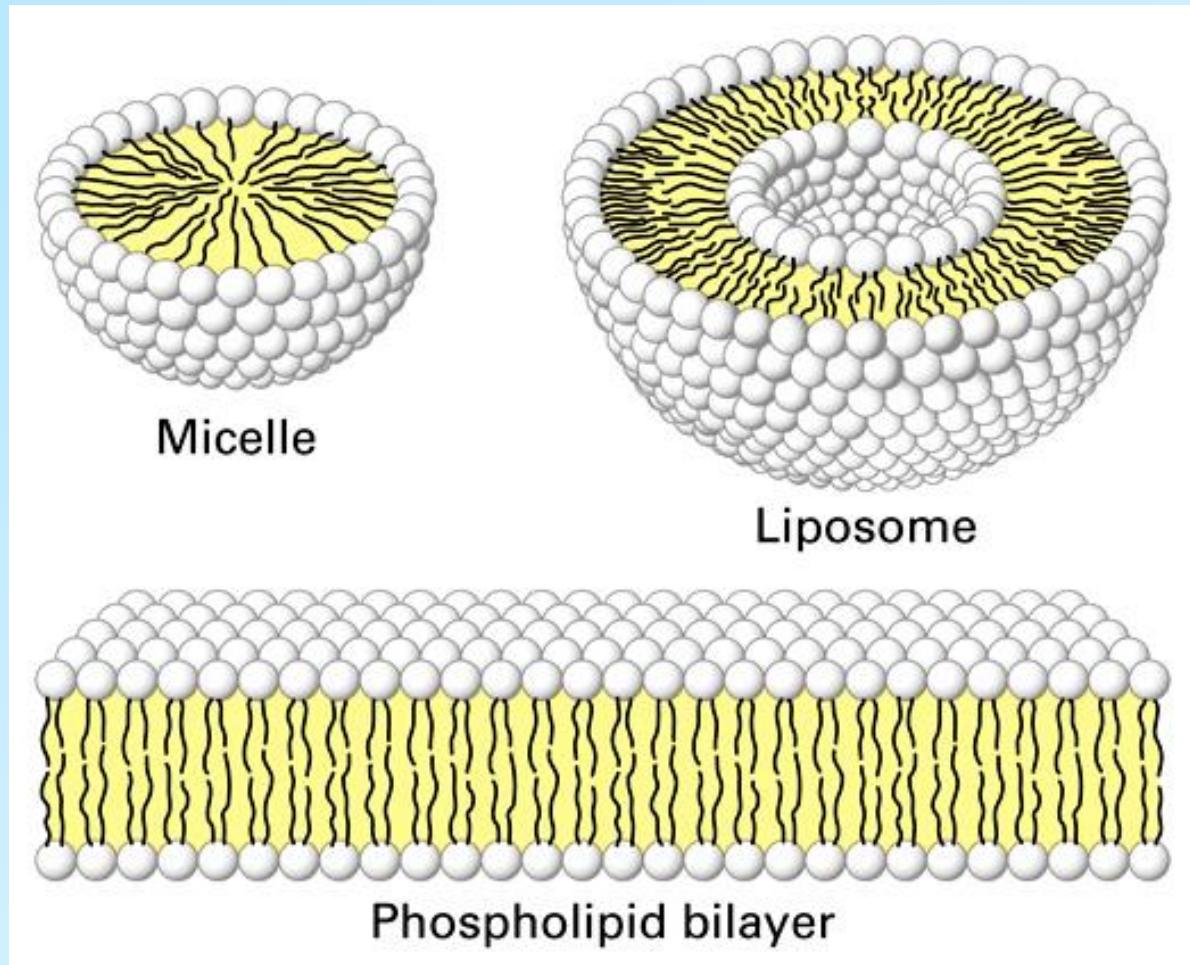
Structure des bulles :



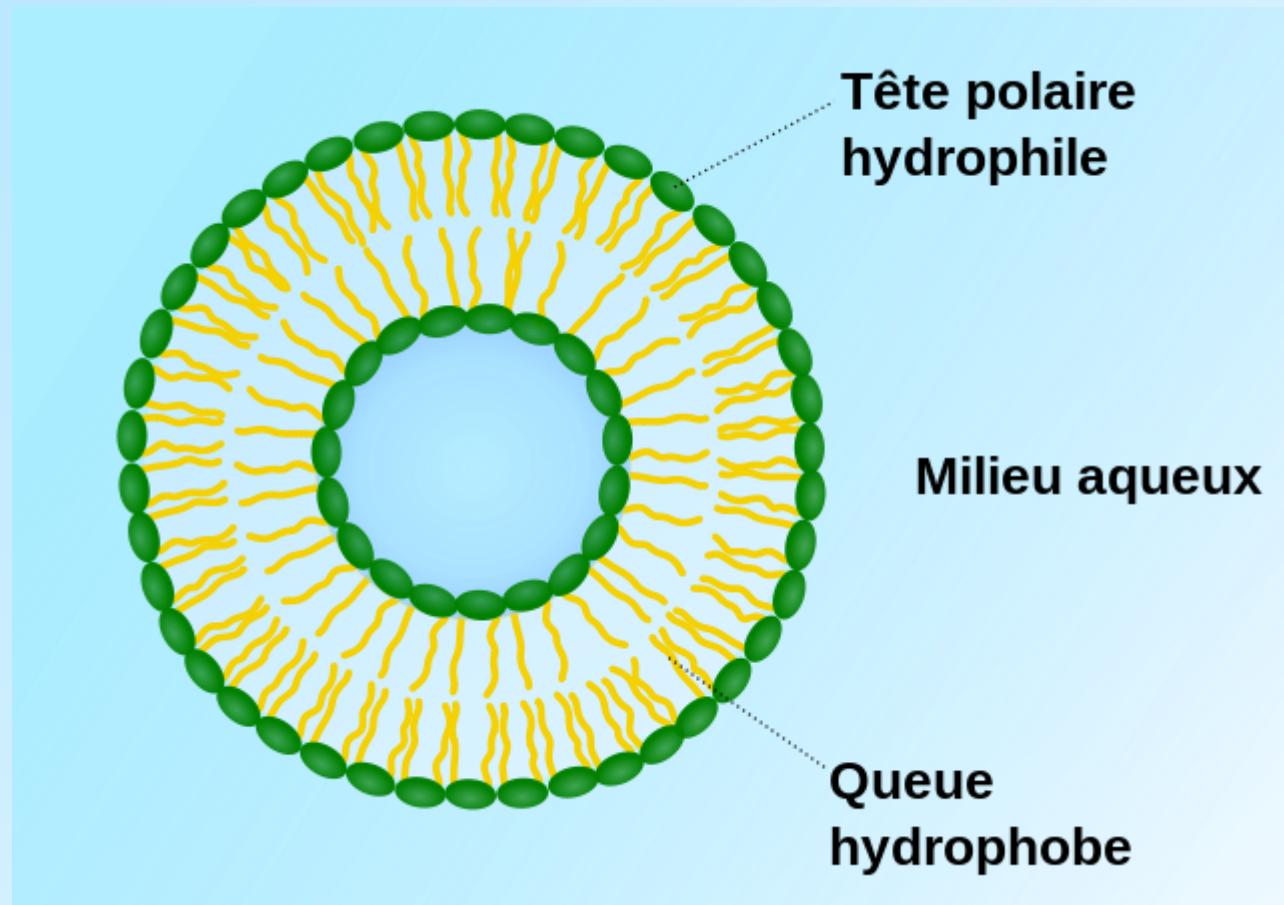
Il se forme une **bicouche** de molécules amphiphiles qui emprisonnent une mince pellicule de liquide.



# *Structures*



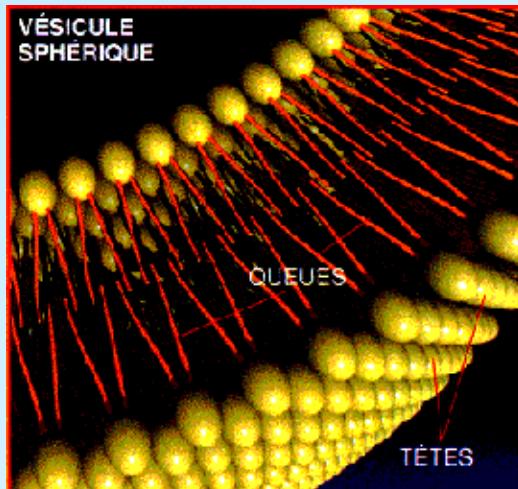
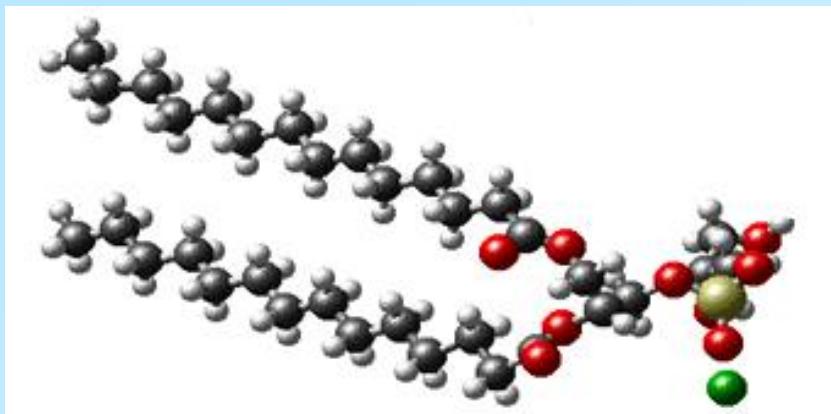
# LIPOSOMES



# Membranes cellulaires

DMPG

sodium 2,3-dimyristoyl-D-glycero-1-phosphate

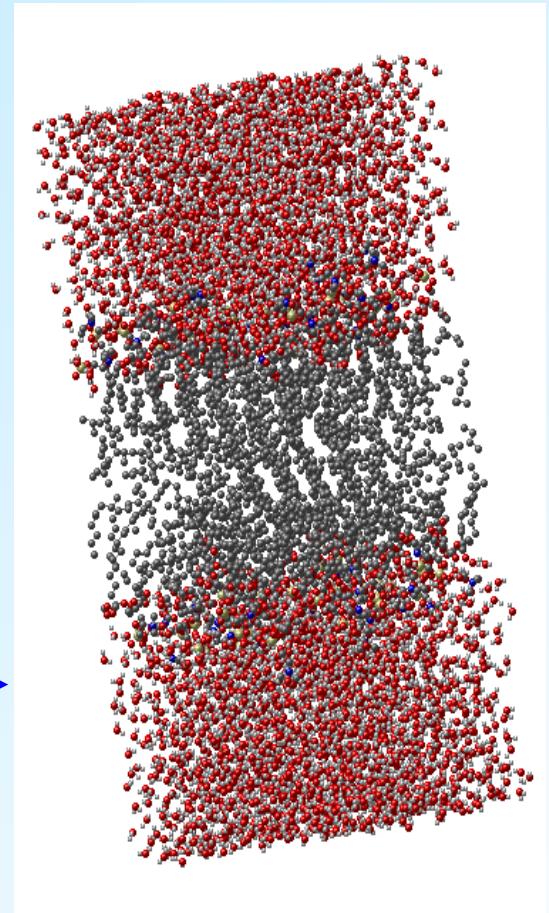


eau →

phospholipides →

eau →

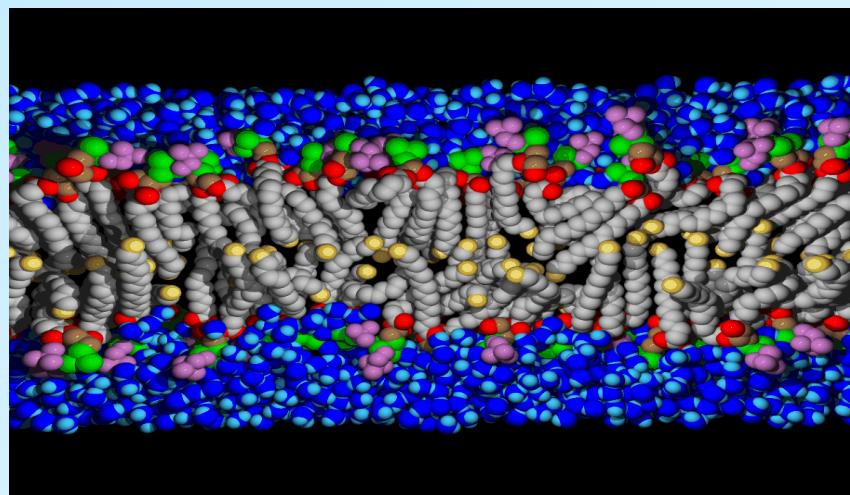
BICOUCHE de phospholipides



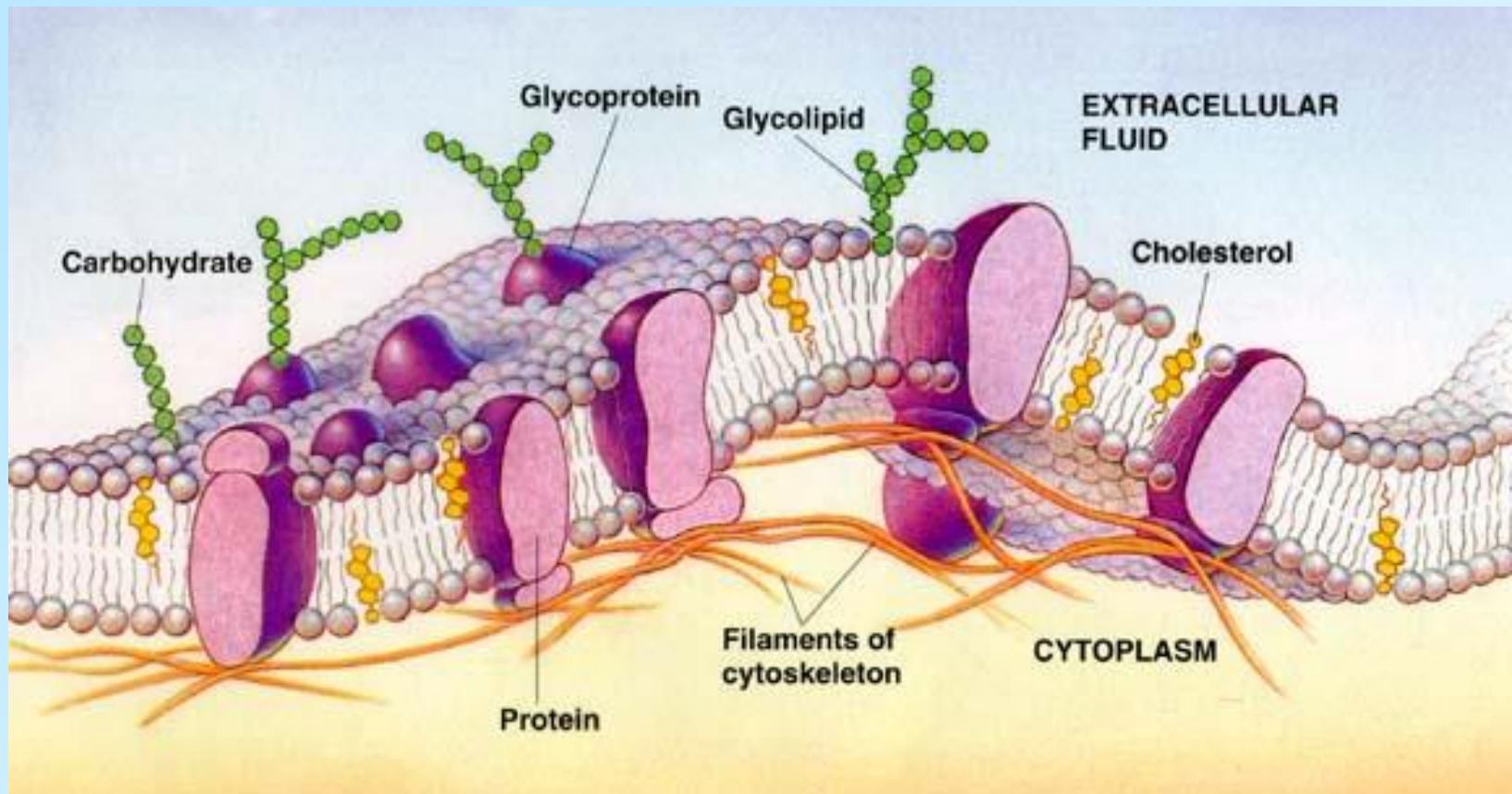
## Vue au microscope électronique



## Simulation moléculaire simplifiée



# Complexité



# Pompe sodium - potassium

